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(The Patent Office will fill in this part)**9812097.5**3. Full name, address and postcode of the or of each applicant (underline all surnames)CeNeS Limited  
St. John's Innovation Centre,  
Cowley Road,  
Cambridge CB4 4WS.

Patents ADP number (if you know it)

If the applicant is a corporate body, give the country/state of incorporation

United Kingdom

709 334 7001

4. Title of the invention Morphine-6-Glucuronide Synthesis

5. Full name, address and postcode in the United Kingdom to which all correspondence relating to this form and translation should be sent

Reddie & Grose  
16 Theobalds Road  
LONDON  
WC1X 8PL

91001

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Country

Priority application  
(if you know it)Date of filing  
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Claim(s)	2
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11. I/We request the grant of a patent on the basis of this application.

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Date

*Reddie & Grose*

4 June 1998

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0171-242 0901

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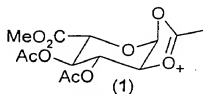
# Morphine-6-Glucuronide Synthesis

The invention provides a novel method for synthesising M6G, and intermediates therefor.

In order to synthesise M6G the major problem to overcome is to obtain the glycoside linkage with very high  $\beta$ -selectivity since prior methods produce the  $\alpha$ -anomer.

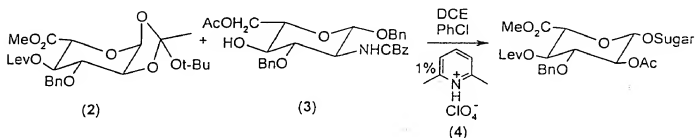
One method for obtaining high  $\beta$ -selectivity is to use trichloroimidate as the leaving group, as shown in WO 93/03051: Figure 1 (Salford Ultrafine Chemicals and Research Limited).

According to the invention, a reaction, which involves the activation of a group on the anomeric position to give the oxonium ion as a reactive intermediate, which also has an acetate group at the C(2) position, proceeds via the cyclic intermediate (1).



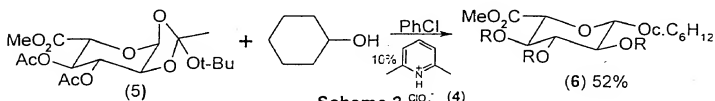
This has an important controlling effect on the stereochemistry on the anomeric site, ensuring that only the  $\beta$ -anomer is obtained in the final product.

Orthoesters are simple to synthesise from their respective bromides<sup>1</sup>. There is a reaction reported in the literature<sup>2</sup> between the glucuronate orthoester (2) and the sugar derivative (3) catalysed by lutidinium perchlorate<sup>3</sup> (4) (Scheme 1).



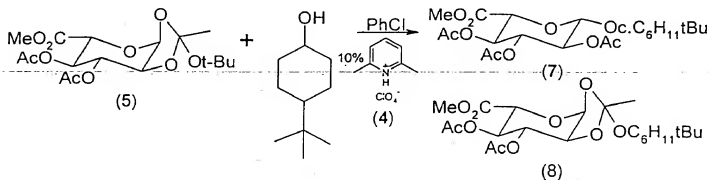
Scheme 1

When this reaction was repeated with the *t*-butyl orthoacetate (5) and cyclohexanol (6 equivalents), the desired product (6) was isolated in 9% yield. Two other products also suggested that they were the desired product, but with the loss of one acetyl group, isolated in a combined yield of 43% (Scheme 2).



Scheme 2

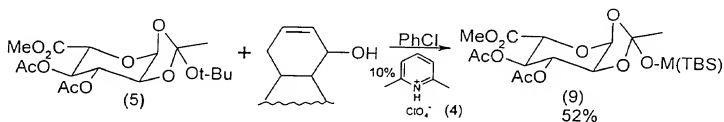
When 1.2 equivalents of 4-*tert*-butylcyclohexanol was used, the desired compound (7) was obtained in 17% yield. Other compounds obtained from the reaction also appeared to contain the desired peaks in the nmr, but after further examination proved to be the product of transorthoesterification (8) (Scheme 3).



Scheme 3

### Reaction of orthoester (5) with protected morphine

Initially, 1.2 equivalents of 3-TBS protected morphine and the orthoester (5) were dissolved in chlorobenzene and half of the solvent was distilled off before 0.1 equivalents of lutidinium perchlorate (4) in chlorobenzene was added. The solvent was continuously distilled off while fresh solvent was added, and after 2.5 h another compound was formed with similar tlc properties to the protected morphine. Workup and chromatography gave a compound which corresponded to trans-orthoesterified material (9). None of the desired material was obtained (Scheme 4).



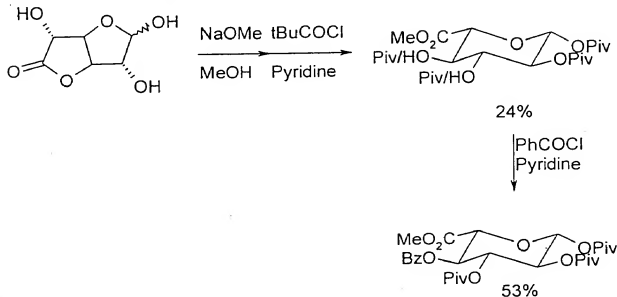
Scheme 4

This product (9) was resubmitted to the reaction conditions (0.1 equivalents of lutidinium perchlorate and protected morphine in refluxing chlorobenzene) with no new products formed after 4h. Two further reactions were attempted using two equivalents of orthoester (5) and 0.2 equivalents of lutidinium perchlorate and 1 equivalent of orthoester (5) and 1.2 equivalents of lutidinium perchlorate, but both gave varying yields of orthoester (9).

We have concluded that a different, more bulky, alkyl group was needed on the orthoester to hinder attack there. Initially, the isopropyl group was examined. However, the initial reaction, perisobutyrylation, failed to give a compound which recrystallised from petrol, so the  $\alpha$  and  $\beta$  anomers could not be separated. Therefore, attention focussed on the pivaloyl group.

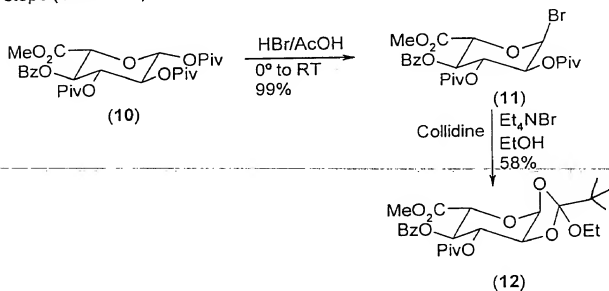
### Synthesis of the ethylorthopivalate (12)

Perpivalation turned out to give a mixture of 3 and 4 non-pivalated material, which was benzoylated (Scheme 5).



Scheme 5

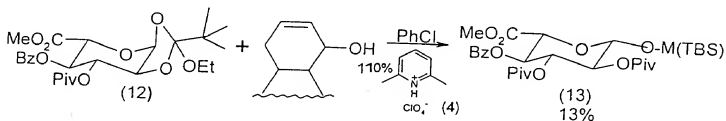
This particular isomer (10) was subjected to bromination (HBr/AcOH) and orthoesterification (Et<sub>4</sub>NBr, collidine, EtOH) to give the orthoester (12) in 57% for 2 steps (Scheme 6).



Scheme 6

### Reaction of orthoester (12) with protected morphine

Subjecting the orthoester (12) to 1.1 equivalents of both TBS protected morphine and lutidinium perchlorate in distilling chlorobenzene for 2h gave another product close to TBS protected morphine on tlc. Workup and purification by chromatography gave a compound (13) which corresponds to protected M6G (with no detectable quantity of  $\alpha$ -anomer) from nmr analysis (Scheme 7).



Scheme 7

A final step in the synthesis of M6G is to hydrolyse the protecting groups from this compound.

This could be achieved in a number of known ways, including that of WO 93/03051.

Steps which might improve the above scheme may include:

- Carrying out the bromination and subsequent reactions on the C(3) and C(4) non-pivalated material (as these unprotected hydroxyls are relatively unreactive). Heating the pivalation reaction to reflux gives a moderate yield of fully pivalated material and it may be desirable to optimise this reaction.
- Attempting to couple the morphine to the brominated material by using a suitably hindered amine base such as *N,N*-di-*iso*-propylethylamine (Hönig's Base). This is based on the observation that on forming the orthoester a small amount of material was formed that appeared to represent direct coupling of ethanol to the anomeric position.
- Changing the base used in the orthoester forming reaction from collidine (which boils at 170°C) to a lower boiling point amine. A catalytic amount of DMAP (4-*N,N*-dimethylaminopyridine) may also be used.
- This reaction could also be done successfully with tosic acid (cheaper and less hazardous) rather than the lutidinium perchlorate (4) used.

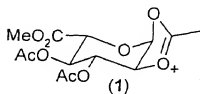


### References

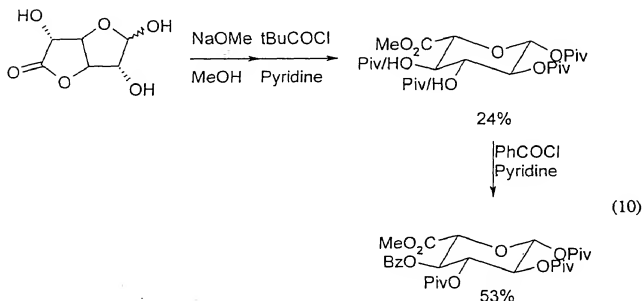
1. For a review of orthoesters and their synthetic applications see N. K. Kochetkov and A. F. Bochkov, *Recent Developments in the Chemistry of Natural Carbon Compounds*, Ed. R. Bognár, V. Bruckner, and Cs. Szántay, Akadémiai Kiadó: Budapest, 1971, vol. 4, p.77-191.
2. H. P. Wessel, L. Labler, and T. B. Tschopp, *Helv. Chim. Acta.*, 1989, 72, 1268.
3. The use of 2,6-dimethylpyridinium perchlorate (4) was first reported by N. K. Kochetkov, A. F. Bochkov, T. A. Sokolovskaya, and V. J. Snyatkova, *Carbohydr. Res.*, 1971, 16, 17.

aims

1. A method for the preferential synthesis of the  $\beta$ -anomer of M6G, which involves the activation of a group on the anomeric position to give the oxonium ion as a reactive intermediate, which also has an acetate group at the C(2) position, and which proceeds via the cyclic intermediate (1).

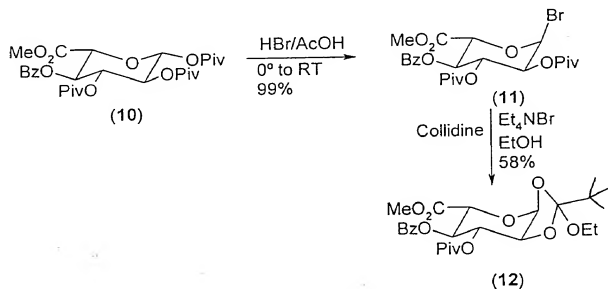


2. Synthesis according to claim 1 which includes the step shown in Scheme 5:



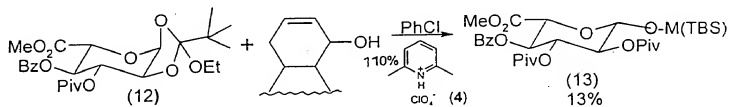
Scheme 5

3. Synthesis according to claim 1 or 2 which includes the step shown in Scheme 6:



Scheme 6

4. Synthesis according to claim 1, 2 or 3 which includes the step shown in scheme 7:



Scheme 7

5. Synthesis according to claim 4 which includes a step to hydrolyse the protecting groups from compound (13).

6. In intermediate compound of formula (1) or a side-group-protected derivative thereof for use in a method of claim 1.

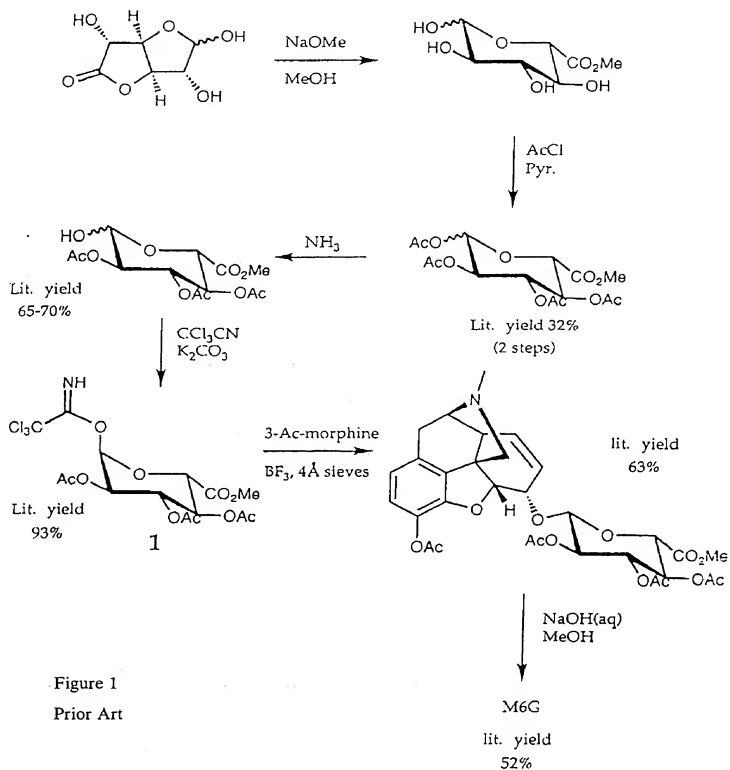


Figure 1  
Prior Art